

## 1,2-Dibenzoylhydrazine

S. Shanmuga Sundara Raj,<sup>a</sup> Bohari M. Yamin,<sup>b</sup>  
Ahmed M. A. Boshala,<sup>c</sup> M. T. H. Tarafder,<sup>c</sup> Karen A.  
Crouse<sup>c</sup> and Hoong-Kun Fun<sup>a\*</sup>

<sup>a</sup>X-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, <sup>b</sup>School of Chemical Sciences and Food Technology, Universiti Kebangsaan Malaysia, 43650 Bangi, Selangor, Malaysia, and <sup>c</sup>Department of Chemistry, Universiti Putra Malaysia, Serdang, Selangor, Malaysia  
Correspondence e-mail: hkfun@usm.my

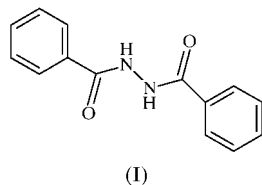
Received 28 April 2000

Accepted 8 May 2000

In the crystal structure of the title compound, C<sub>14</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub>, the molecule lies about a twofold axis; two carbonyl groups and the H atoms of the N—N bond are in a *trans* orientation with respect to each other. In the crystal, each molecule is linked to the other and *vice versa* by intermolecular N—H···O hydrogen bonds between the amide hydrogen and the O atoms of neighbouring molecules to form two ten-membered rings, each of which has the graph-set motif C4R<sub>2</sub><sup>2</sup>(10). This extends as a polymeric chain along the *c* axis.

## Comment

Both *S*-methylthiocarbazate and *S*-benzylcarbazate can be prepared by the addition of chloromethane or benzyl chloride to a mixture of hydrazine and carbon disulfide, respectively (Lanfredi *et al.*, 1977; Mattes & Weber, 1980; Ali & Tarafder, 1977). However, under the same conditions, 1,2-dibenzoylhydrazine, (I), was obtained instead of *S*-benzoyldithiocarbazate when benzoyl chloride was added to the reaction mixture.



The dihedral angle of 24.0 (1)° between the phenyl ring and the N1—C7=O1 group shows that the molecule is not planar. The asymmetric unit contains half the molecule and the other half is related by a crystallographic twofold axis passing through the N1—N1<sup>i</sup> bond [symmetry code: (i)  $-x, y, \frac{1}{2} - z$ ]. The symmetry-related group adopts an anticlinical orientation with respect to the N—N bond, the torsion angle C7—N1—N1<sup>i</sup>—C7<sup>i</sup> being 104.3 (2)°. The two carbonyl groups and the H atoms of the N—N bond are in a *trans* orientation with respect to each other.

The bond lengths and angles are comparable with those of *N,N'*-bis(picolinoyl)hydrazine (Shao *et al.*, 1999), except for C=O which is slightly longer and the N—C—C angle [114.7 (1)°] which is 2° larger, possibly due to the greater twist along the N—N<sup>i</sup> bond [torsion angle C—N—N<sup>i</sup>—C<sup>i</sup> in Shao *et al.* (1999) is  $-70.9$  (2)°].

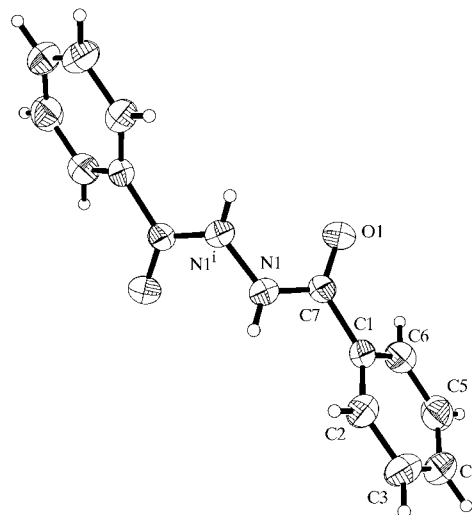


Figure 1

The structure of the title compound showing 50% probability displacement ellipsoids and the atom-numbering scheme [symmetry code: (i)  $-x, y, \frac{1}{2} - z$ ].

In the crystal, each molecule is linked to another and *vice versa* by intermolecular N—H···O hydrogen bonds between the amide H atom and the O atoms of neighbouring molecules to form two ten-membered rings, each of which has the graph-set motif C4R<sub>2</sub><sup>2</sup>(10) (Bernstein *et al.*, 1995). This extends as a polymeric chain along the *c* axis.

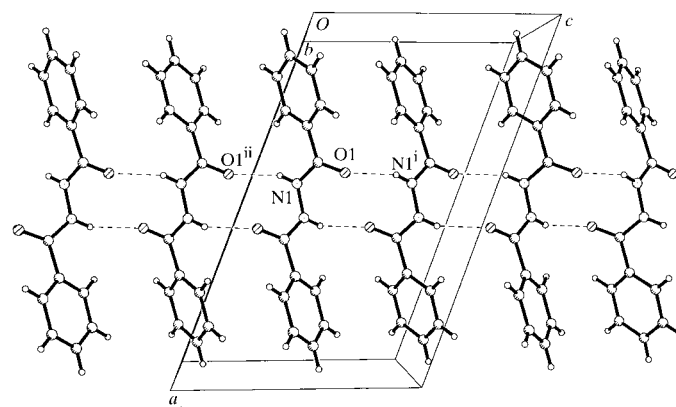


Figure 2

The polymeric chain of the packing of the title compound along the *c* axis.

## Experimental

The title compound was prepared by adding benzoyl chloride to a mixture of hydrazine, potassium hydroxide and carbon disulfide in 20% aqueous ethanol at 273 K. The precipitate was washed with water and dried. The crystal used for analysis was obtained by recrystallization from ethanol.

## Crystal data

$C_{14}H_{12}N_2O_2$	$D_x = 1.339 \text{ Mg m}^{-3}$
$M_r = 240.26$	Mo $K\alpha$ radiation
Monoclinic, $C2/c$	Cell parameters from 2038 reflections
$a = 14.5439 (2) \text{ \AA}$	$\theta = 3.00\text{--}28.29^\circ$
$b = 9.7314 (3) \text{ \AA}$	$\mu = 0.092 \text{ mm}^{-1}$
$c = 9.0181 (3) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 110.938 (1)^\circ$	Block, light yellow
$V = 1192.07 (6) \text{ \AA}^3$	$0.32 \times 0.30 \times 0.28 \text{ mm}$
$Z = 4$	

## Data collection

Siemens SMART CCD area-detector diffractometer	$R_{\text{int}} = 0.054$
$\omega$ scans	$\theta_{\text{max}} = 28.30^\circ$
4148 measured reflections	$h = -19 \rightarrow 14$
1471 independent reflections	$k = -12 \rightarrow 12$
965 reflections with $I > 2\sigma(I)$	$l = -10 \rightarrow 12$

## Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0975P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.053$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.164$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 0.959$	$\Delta\rho_{\text{max}} = 0.30 \text{ e \AA}^{-3}$
1471 reflections	$\Delta\rho_{\text{min}} = -0.28 \text{ e \AA}^{-3}$
83 parameters	Extinction correction: <i>SHELXTL</i>
H-atom parameters constrained	(Sheldrick, 1997)
	Extinction coefficient: 0.021 (5)

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

O1—C7	1.229 (2)	N1—N1 <sup>i</sup>	1.385 (3)
N1—C7	1.354 (2)		
C7—N1—N1 <sup>i</sup>	118.6 (1)	O1—C7—C1	122.0 (2)
O1—C7—N1	121.3 (2)	N1—C7—C1	116.7 (1)

Symmetry code: (i)  $-x, y, \frac{1}{2} - z$ .

**Table 2**

Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N1—H1A $\cdots$ O1 <sup>i</sup>	0.86	2.13	2.924 (2)	154

Symmetry code: (i)  $x, 2 - y, \frac{1}{2} + z$ .

Data collection was as described in Shanmuga Sundara Raj *et al.* (1999).

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT*; data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PARST* (Nardelli, 1995).

The authors would like to thank the Malaysian Government, Universiti Kebangsaan Malaysia, Universiti Putra Malaysia and Universiti Sains Malaysia for research grant R&D Nos. 03-02-02-005 and 305/pfizik/610942. SSSR thanks the Universiti Sains Malaysia for a Visiting Postdoctoral Fellowship.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: NA1474). Services for accessing these data are described at the back of the journal.

## References

- Ali, M. A. & Tarafder, M. T. H. (1977). *J. Inorg. Nucl. Chem.* **39**, 1785–1788.
- Bernstein, J., Davis, R. E., Shimon, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Lanfredi, A. M. M., Tiripiccio, A., Camellini, M. T., Monaci, A. & Tarli, S. (1977). *J. Chem. Soc. Dalton Trans.* pp. 417–422.
- Mattes, R. & Weber, H. (1980). *J. Chem. Soc. Dalton Trans.* pp. 423–425.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Shanmuga Sundara Raj, S., Renganayaki, S., Subramanian, E. & Fun, H.-K. (1999). *Acta Cryst.* **C55**, 2182–2184.
- Shao, S.-C., Zhu, D.-R., Song, Y., You, X.-Z., Shanmuga Sundara Raj, S. & Fun, H.-K. (1999). *Acta Cryst.* **C55**, 1841–1843.
- Sheldrick, G. M. (1997). *SHELXTL*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.